

1. AgF (s)
2. AgNO<sub>3</sub>, 20° (s)
3. AgNO<sub>3</sub>, razne temperature (s)
4. AlBr<sub>3</sub> (s)
5. AlCl<sub>3</sub>, 15° (s)
6. AlCl<sub>3</sub>, 18° (s)
7. Al(NO<sub>3</sub>)<sub>3</sub> (s)
8. Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> (s)
9. AlK(SO<sub>4</sub>)<sub>2</sub> (s)
10. AlNH<sub>4</sub>(SO<sub>4</sub>)<sub>2</sub> (s)
11. AuCl<sub>3</sub> (s)
12. BaBr<sub>2</sub> (s)
13. Ba(CH<sub>3</sub>COO)<sub>2</sub> (s) *acetate*
14. BaCl<sub>2</sub> (s)
15. BaCl<sub>2</sub>, razne temperature (s)
16. BaI<sub>2</sub> (s)
17. Ba(NO<sub>3</sub>)<sub>2</sub> (s)
18. BeCl<sub>2</sub> (s)
19. Be(NO<sub>3</sub>)<sub>2</sub> (s)
20. BeSO<sub>4</sub>
21. CaBr<sub>2</sub> (s)
22. CaBr<sub>2</sub>, razne temperature (s)
23. Ca(CHO<sub>2</sub>)<sub>2</sub> (s) *formate*
24. Ca(CH<sub>3</sub>COO)<sub>2</sub> (s) *acetate*
25. CaCl<sub>2</sub> (s)
26. CaI<sub>2</sub> (s)
27. CaI<sub>2</sub>, razne temperature (s)
28. Ca(NO<sub>3</sub>)<sub>2</sub> (s)
29. Ca(OH)<sub>2</sub> (s)
30. CdBr<sub>2</sub> (s)
31. CdCl<sub>2</sub> (s)
32. CdI<sub>2</sub> (s)
33. Cd(NO<sub>3</sub>)<sub>2</sub> (s)
34. CdSO<sub>4</sub> (s)
35. CoBr<sub>2</sub> (s)
36. CoCl<sub>2</sub> (s)
37. Cd(NO<sub>3</sub>)<sub>2</sub> (s)
38. CoSO<sub>4</sub>, razne temperature (s)
39. CrBr<sub>3</sub> (s)
40. CrCl<sub>3</sub> (s), dvije boje kristala (s)
41. Cr(NO<sub>3</sub>)<sub>3</sub> (s)
42. CrO<sub>3</sub> (s)
43. Cr<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> (s)
44. CrK(SO<sub>4</sub>)<sub>2</sub> ljubičasti (s)
45. CrK(SO<sub>4</sub>)<sub>2</sub> zeleni (s)

46.  $\text{CrNH}_4(\text{SO}_4)_2 \times 12\text{H}_2\text{O}$  ljubičasti (s)  
47.  $\text{CrNH}_4(\text{SO}_4)_2 \times 12\text{H}_2\text{O}$  zeleni (s)  
48. CsBr (s)  
49. CsCl (s)  
50. CsI (s)  
51. CsNO<sub>3</sub> (s)  
52. Cs<sub>2</sub>SO<sub>4</sub> (s)  
53. CuBr<sub>2</sub> (s)  
54. Cu(CH<sub>3</sub>COO)<sub>2</sub> (s) *acetate*  
55. CuCl<sub>2</sub> (s)  
56. Cu(NO<sub>3</sub>)<sub>2</sub> (s)  
57. CuSO<sub>4</sub> (s)  
58. FeCl<sub>2</sub> (s)  
59. FeCl<sub>3</sub> (s)  
60. FeK(SO<sub>4</sub>)<sub>2</sub> (s)  
61. FeNH<sub>4</sub>(SO<sub>4</sub>)<sub>2</sub> (s)  
62. Fe(NO<sub>3</sub>)<sub>2</sub> (s)  
63. FeSO<sub>4</sub> (s)  
64. Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> (s)  
65. HgCl<sub>2</sub> (s)  
66. H<sub>3</sub>AsO<sub>4</sub> (s)  
67. H<sub>3</sub>BO<sub>3</sub> (s)  
68. HBr (g)  
69. HCN (g)  
70. HCl, 15° (g)  
71. HCl, 20° (g)  
72. HClO<sub>3</sub> (?)  
73. HClO<sub>4</sub>, 15° (l)  
**74. HClO<sub>4</sub>, 20° (l)**  
75. HF (g)  
76. HI (g)  
77. HIO<sub>3</sub> (?)  
78. HIO<sub>4</sub> (?)  
**79. HNO<sub>3</sub>, 15° (l)**  
**80. HNO<sub>3</sub>, 20° (l)**  
**81. otopine NO<sub>2</sub> (g) u HNO<sub>3</sub>(l)**  
**82. otopine NO<sub>2</sub> (g) u HNO<sub>3</sub>(aq)**  
**83. otopine N<sub>2</sub>O<sub>3</sub> (g) u H<sub>2</sub>SO<sub>4</sub>(aq)**  
**84. H<sub>2</sub>O<sub>2</sub> (l)**  
**85. H<sub>3</sub>PO<sub>4</sub> (l)**  
**86. H<sub>2</sub>SO<sub>4</sub> 15° (l)**  
**87. H<sub>2</sub>SO<sub>4</sub> 20° (l)**  
**88. H<sub>2</sub>SO<sub>4</sub> više temperature (l)**  
**89. oleum, otopine SO<sub>3</sub> (g) u H<sub>2</sub>SO<sub>4</sub>(l)**  
**90. oleum, preračunato na monohidrat**

91.  $\text{H}_2\text{S}_2\text{O}_8$  (?)  
92.  $\text{H}_2\text{SeO}_4$  (?)  
93.  $\text{H}_2\text{SiF}_6$  (?)  
94.  $\text{InBr}_3$  (s)  
95.  $\text{KBr}$  (s)  
96.  $\text{KBrO}_3$  (s)  
97.  $\text{KCH}_3\text{COO}$  (s) *acetate*  
98.  $\text{K}_2\text{C}_4\text{H}_4\text{O}_6$  (s) *tartrate*  
99.  $\text{KCN}$  (s)  
100.  $\text{K}_2\text{CO}_3$  (s)  
**101.**  $\text{K}_2\text{CO}_3$ , razne temperature (s)  
102.  $\text{K}_2\text{C}_2\text{O}_4$  (s) *oxalate*  
103.  $\text{KCl}$  (s)  
**104.**  $\text{KCl}$ , razne temperature (s)  
105.  $\text{KClO}_3$  (s)  
106.  $\text{KClO}_4$  (s)  
107.  $\text{K}_2\text{CrO}_4$  (s)  
108.  $\text{K}_2\text{Cr}_2\text{O}_7$  (s)  
109.  $\text{KF}$  (s)  
110.  $\text{K}_3[\text{Fe}(\text{CN})_6]$  (s)  
111.  $\text{K}_4[\text{Fe}(\text{CN})_6]$  (s)  
112.  $\text{KHCO}_3$  (s)  
113.  $\text{KH}_2\text{PO}_4$  (s)  
114.  $\text{KHS}$  (s)  
115.  $\text{KHSO}_4$  (s)  
116.  $\text{KI}$  (s)  
117.  $\text{KIO}_3$  (s)  
118.  $\text{KMnO}_4$  (s)  
119.  $\text{K}_2\text{MoO}_4$  (s)  
120.  $\text{KN}_3$  (s)  
121.  $\text{KNO}_2$  (s)  
122.  $\text{KNO}_3$  (s)  
**123.**  $\text{KNO}_3$ , razne temperature (s)  
**124.**  $\text{KOH}$ , 15° (s)  
**125.**  $\text{KOH}$ , 20° (s)  
126.  $\text{K}_2\text{S}$  (s)  
127.  $\text{KSCN}$  (s)  
128.  $\text{K}_2\text{SO}_3$  (s)  
129.  $\text{K}_2\text{SO}_4$  (s)  
130.  $\text{K}_2\text{SiO}_3$  (s)  
131.  $\text{K}_2\text{WO}_4$  (s)  
132.  $\text{La}(\text{NO}_3)_3$  (s)  
133.  $\text{LiBr}$  (s)  
134.  $\text{LiCl}$  (s)  
135.  $\text{LiI}$  (s)

136. LiNO<sub>3</sub> (s)  
137. LiOH (s)  
138. Li<sub>2</sub>SO<sub>4</sub> (s)  
139. MgBr<sub>2</sub> (s)  
140. Mg(CHOO)<sub>2</sub> (s) *formate*  
141. Mg(CH<sub>3</sub>COO)<sub>2</sub> (s) *acetate*  
142. MgCl<sub>2</sub>, 20°(s)  
143. MgCl<sub>2</sub>, 30°(s)  
144. MgCrO<sub>4</sub> (s)  
145. MgI<sub>2</sub> (s)  
146. Mg(NO<sub>3</sub>)<sub>2</sub> (s)  
147. MgSO<sub>4</sub> (s)  
148. MnBr<sub>2</sub> (s)  
149. MnCl<sub>2</sub> (s)  
150. Mn(NO<sub>3</sub>)<sub>2</sub> (s)  
151. MnSO<sub>4</sub> (s)  
**152.** NH<sub>3</sub>, 15° (g)  
**153.** NH<sub>3</sub>, 20° (g)  
154. NH<sub>2</sub>OH (s)  
155. NH<sub>2</sub>OH×HCl (s)  
156. NH<sub>4</sub>Br (s)  
157. NH<sub>4</sub>CHOO (s) *formate*  
158. NH<sub>4</sub>CH<sub>3</sub>COO (s) *acetate*  
159. (NH<sub>4</sub>)<sub>2</sub>CO<sub>3</sub> (s)  
160. NH<sub>4</sub>Cl, 20° (s)  
**161.** NH<sub>4</sub>Cl, razne temperature (s)  
162. NH<sub>4</sub>F (s)  
163. NH<sub>4</sub>I (s)  
164. NH<sub>4</sub>NO<sub>3</sub> (s)  
**165.** NH<sub>4</sub>NO<sub>3</sub>, razne temperature (s)  
166. NH<sub>4</sub>SCN (s)  
167. (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> (s)  
168. N<sub>2</sub>H<sub>4</sub> (l)  
169. N<sub>2</sub>H<sub>4</sub>×2HCl (s)  
170. Na<sub>3</sub>AsO<sub>4</sub> (s)  
171. Na<sub>2</sub>B<sub>2</sub>O<sub>4</sub> (s)  
172. Na<sub>2</sub>B<sub>4</sub>O<sub>7</sub> (s)  
173. NaBr (s)  
174. NaBrO<sub>3</sub> (s)  
175. NaCHOO (s) *formate*  
176. NaCH<sub>3</sub>COO (s) *acetate*  
177. Na<sub>2</sub>C<sub>4</sub>H<sub>4</sub>O<sub>6</sub> (s) *tartrate*  
178. NaC<sub>4</sub>H<sub>5</sub>O<sub>6</sub> (s) *hydrogentartrate*  
179. NaC<sub>4</sub>H<sub>5</sub>O<sub>5</sub> (s) *malate*  
180. Na<sub>2</sub>CO<sub>3</sub>, 20° (s)c2

181.  $\text{Na}_2\text{CO}_3$ , 30° (s)  
182.  $\text{Na}_2\text{C}_2\text{O}_4$ , (s) *oxalate*  
183.  $\text{NaCl}$ , 15° (s)  
184.  $\text{NaCl}$ , 20° (s)  
**185.**  $\text{NaCl}$ , razne temperature (s)  
186.  $\text{NaClO}$  (s)  
187.  $\text{NaClO}_3$  (s)  
188.  $\text{NaClO}_4$  (s)  
189.  $\text{Na}_2\text{CrO}_4$  (s)  
190.  $\text{Na}_2\text{Cr}_2\text{O}_7$  (s)  
191.  $\text{NaF}$  (s)  
192.  $\text{Na}_2\text{HAsO}_4$  (s)  
193.  $\text{NaHCO}_3$  (s)  
194.  $\text{Na}_2\text{HPO}_4$  (s)  
195.  $\text{NaH}_2\text{PO}_4$  (s)  
196.  $\text{NaHSO}_3$  (s)  
197.  $\text{NaHSO}_4$  (s)  
198.  $\text{NaI}$  (s)  
199.  $\text{NaKC}_4\text{H}_4\text{O}_6$  *tartrate*  
200.  $\text{Na}_2\text{MoO}_4$  (s)  
201.  $\text{NaN}_3$  (s)  
202.  $\text{NaNO}_2$  (s)  
203.  $\text{NaNO}_3$  (s)  
**204.**  $\text{NaOH}$ , 20° (s)  
205.  $\text{Na}_2\text{O} \times 1.69 \text{ SiO}_2$  (s) *waterglass*  
206.  $\text{Na}_2\text{O} \times 2.06 \text{ SiO}_2$  (s) *waterglass*  
207.  $\text{Na}_2\text{O} \times 2.4 \text{ SiO}_2$  (s) *waterglass*  
208.  $\text{Na}_2\text{O} \times 3.36 \text{ SiO}_2$  (s) *waterglass*  
209.  $\text{Na}_3\text{PO}_4$  (s)  
210.  $\text{Na}_4\text{P}_2\text{O}_7$  (s)  
211.  $\text{Na}_2\text{S}$  (s)  
212.  $\text{NaSCN}$  (s)  
213.  $\text{Na}_2\text{SO}_3$  (s)  
214.  $\text{Na}_2\text{SO}_4$ , 10°(s)  
215.  $\text{Na}_2\text{SO}_4$ , 20°(s)  
216.  $\text{Na}_2\text{SO}_4$ , 25°(s)  
217.  $\text{Na}_2\text{S}_2\text{O}_3$  (s)  
218.  $\text{Na}_2\text{SiO}_3$  (s)  
219.  $\text{Na}_2\text{SnO}_3$  (s)  
220.  $\text{Na}_2\text{WO}_4$  (s)  
221.  $\text{NiBr}_2$  (s)  
222.  $\text{NiCl}_2$  (s)  
223.  $\text{Ni}(\text{NO}_3)_2$  (s)  
224.  $\text{NiSO}_4$  (s)  
225.  $\text{Pb}(\text{CH}_3\text{COO})_2$  (s) *acetate*

226.  $\text{PbCl}_2$  (s)  
227.  $\text{Pb}(\text{NO}_3)_2$  (s)  
228.  $\text{RbBr}$  (s)  
229.  $\text{RbCl}$  (s)  
230.  $\text{RbI}$  (s)  
231.  $\text{RbNO}_3$  (s)  
232.  $\text{RbOH}$  (s)  
233.  $\text{Rb}_2(\text{SO}_4)$  (s)  
234.  $\text{SO}_2$  (g)  
235.  $\text{SnCl}_2$  (s)  
236.  $\text{SnCl}_4$ , 15° (s)  
237.  $\text{SnCl}_4$ , 18° (s)  
238.  $\text{SrBr}_2$  (s)  
239.  $\text{SrCl}_2$  (s)  
240.  $\text{SrI}_2$  (s)  
241.  $\text{Sr}(\text{NO}_3)_2$  (s)  
242.  $\text{Sr}(\text{OH})_2$  (s)  
243.  $\text{Th}(\text{NO}_3)_4$  (s)  
244.  $\text{Tl}(\text{NO}_3)_2$  (s)  
245.  $\text{Tl}_2\text{SO}_4$  (s)  
246.  $\text{UO}_2(\text{CH}_3\text{COO})_2$  (s) *uranyl acetate*  
247.  $\text{UO}_2(\text{NO}_3)_2$ , 17° (s) *uranyl*  
248.  $\text{UO}_2(\text{NO}_3)_2$ , 25° (s) *uranyl*  
249.  $\text{ZnBr}_2$  (s)  
250.  $\text{ZnCl}_2$  (s)  
251.  $\text{ZnI}_2$  (s)  
252.  $\text{Zn}(\text{NO}_3)_2$  (s)  
253.  $\text{ZnSO}_4$  (s)